

Appl. No. : **09/913,799**
Filed : **December 31, 2001**

REMARKS

Claim 1 has been amended by incorporating the limitation of Claims 2 and 4. Claim 14 has been amended by incorporating the limitation of Claim 17. Claims 2, 4 and 17 have been canceled without prejudice. Claims 3, 5, 8, 12 and 18-21 have been amended to correct clerical errors or dependencies. Accordingly, Claims 1, 3, 5-16 and 18-22 are pending in this application. As such, no new matter has been added. Applicant respectfully requests entry of the amendments and reconsideration of the application in view of the amendments and the following remarks.

Rejection of Claims 1 and 14 Under 35 U.S.C. § 102

Claims 1 and 14 have been rejected under 35 U.S.C. § 102(b) as being anticipated by Shibata Teruhiko JP 60-255729. Applicant respectfully traverses the rejection.

Claim 1 has been amended by incorporating the limitation of Claims 2 and 4 which are not rejected under § 102 by the Examiner. Claim 14 has been amended by incorporating the limitation of Claim 17 which is not rejected under § 102 by the Examiner. Thus, these rejections are moot.

Rejection of Claims 2, 3, 7-13, 16, 21 and 22 Under 35 U.S.C. § 103

Claims 2, 3, 7-13, 16, 21 and 22 have been rejected under 35 U.S.C. § 103(a) as being unpatentable over Shibata Teruhiko JP 60-255729 or Sachiko JP 09-290260 in combination with Iwata further in view of Sommerville et al. Applicant respectfully traverses the rejection. Claim 2 has been canceled without prejudice.

Claim 1 has been amended by incorporating the limitation of Claim 2 and Claim 4 which is not rejected based on the above references by the Examiner. Claim 14 has been amended by incorporating the limitation of Claim 17 which is not rejected based on the above references by the Examiner.

Claims 3 and 7-13 are ultimately dependent on Claim 1 which is not applicable to this rejection. Claims 16, 21 and 22 are ultimately dependent on Claim 14 which is not applicable to this rejection. Thus, these rejections are moot.

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Rejection of Claims 4-6 and 17-19 Under 35 U.S.C. § 103

Claims 4-6 and 17-19 have been rejected under 35 U.S.C. § 103(a) as being unpatentable over Shibata Teruhiko JP 60-255729 or Sachiko JP 09-290260 in combination with Iwata further in view of Remesy WO99/40801. Applicant respectfully traverses the rejection. Claims 4 and 17 have been canceled without prejudice. Claims 5 and 6 are ultimately dependent on Claim 1 as amended herein. Claims 18 and 19 are ultimately dependent on Claim 14 as amended herein.

The Examiner has admitted in the Office action that none of the references teach that the mineral components added to the water are magnesium and calcium and that the weight ratio of the magnesium to calcium (Mg/Ca) is adjusted having a ration of 4/1 to 1/3.

On the other hand, the Examiner has asserted that Remesy WO99/40801 teaches a beverage such as water, which is supplemented with calcium, and magnesium wherein the Ca/Mg weight ration is ranged between 1 and 6 which provides a range, which overlaps with applicant's range.

This application is the U.S. National Phase under 35 U.S.C. § 371 of International Application PCT/JP99/07095, filed December 17, 1999, which claims priority based on JP 11-35145, filed February 15, 1999. Applicant has submitted a copy of verified English translation of JP 11-35145 herewith. The priority document discloses the currently claimed invention.

It is noted that Remesy WO99/40801 was published on August 19, 1999 which is later than the filing date of the JP 11-35145, i.e., February 15, 1999. Accordingly, Remesy WO99/40801 does not serve as prior art. Thus, this rejection is moot. Applicant respectfully requests withdrawal of this rejection.

CONCLUSION

In light of the Applicant's amendments to the claims and the foregoing Remarks, it is respectfully submitted that the present application is in condition for allowance. Should the Examiner have any remaining concerns which might prevent the prompt allowance of the

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application, the Examiner is respectfully invited to contact the undersigned at the telephone number appearing below.

Please charge any additional fees, including any fees for additional extension of time, or credit overpayment to Deposit Account No. 11-1410. A duplicate copy of this sheet is enclosed.

Respectfully submitted,

KNOBBE, MARTENS, OLSON & BEAR, LLP

Dated: January 26, 2004

By:



Katsuhiro Arai
Registration No. 43,315
Agent of Record
Customer No. 20,995
(949) 760-0404

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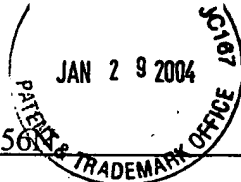


VERIFICATION OF TRANSLATION

I, Tomohisa Sugiyama of Knobbe, Martens, Olson & Bear, LLP, am the translator of the English language document attached and I state that the attached document is a true translation of a Certified Copy of Japanese Patent Application No. 11-035145 filed on February 15, 1999.

Dated this 7th day of January, 2004

Signature of Translator Tomohisa Sugiyama



Filing Date: February 15, 1999

Reference Number = KOAK956N

Patent Application No. 11-035145 Page 1/2

[Document] Patent Application
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[Filing Date] February 15, 1999
[Addressed to] Director of Patent Office Kenzi Isayama
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[Inventor]
 [Address or Domicile] AKO KASEI CO., LTD.,
 329-banchi, Sagoshi, Ako-city, Hyogo, Japan
 [Name] Yoshinari Ikegami
[Inventor]
 [Address or Domicile] AKO KASEI CO., LTD.,
 329-banchi, Sagoshi, Ako-city, Hyogo, Japan
 [Name] Hiroshi Tabuchi
[Inventor]
 [Address or Domicile] AKO KASEI CO., LTD.,
 329-banchi, Sagoshi, Ako-city, Hyogo, Japan
 [Name] Koji Nakagawa
[Applicant]
 [Identification Number] 592015802
 [Address or Domicile] 329-banchi, Sagoshi, Ako-city, Hyogo, Japan
 [Name] AKO KASEI CO., LTD.
 [Representative] Michio Ishino
[Agent]
 [Identification Number] 100105061
 [Patent Attorney]
 [Name] Yoshihiro Kodama
[Identification of Fees]
 [Advanced Payment Ledger Number] 056845
 [Advanced Payment] 21,000 yen
[Priority Right Claim based on Submitted Application]
 [Application Number] Patent Application Number 51308 in 1998
 [Application Date] February 18, 1998

[Priority Right Claim based on Submitted Application]

[Application Number] Patent Application Number 32760 in 1999

[Application Date] February 10, 1999

[Priority Right Claim based on Submitted Application]

[Application Number] Patent Application Number 34162 in 1999

[Application Date] February 12, 1999

[List of Submitted Documents]

[Document] Specification 1

[Document] Abstract 1

[Comprehensive Power of Attorney Number] 9703120

[Proof Required] Yes

[Name of Document]
[Title of the Invention]
[Claims]

Specification
Drink Using Seawater and Method for Producing the Same

[Claim 1] A drink, which is characterized in that water-soluble mineral components are added to water obtained by desalting seawater.

[Claim 2] The drink as claimed in Claim 1, which is characterized in that said water-soluble mineral components are obtained from seawater.

[Claim 3] The drink as claimed in Claim 1 or 2, which is characterized in that water-soluble mineral components obtained from seawater are added to water obtained by desalting the same seawater.

[Claim 4] The drink as claimed in any one of Claims 1 to 3, which is characterized in that mineral components obtained from seawater are magnesium and calcium.

[Claim 5] The drink as claimed in any one of Claims 1 to 4, which is characterized in that the weight ratio of magnesium to calcium (Mg/Ca) is adjusted to 4/1 to 1/3.

[Claim 6] The drink as claimed in Claim 5, for which is characterized in that the weight ratio of magnesium to calcium (Mg/Ca) is adjusted to 3/1.

[Claim 7] The drink as claimed in any one of Claims 1 to 6, which is characterized in that hardness of water measured by the EDTA method is from 100 to 3,000.

[Claim 8] The drink as claimed in Claim 7, which is characterized in that hardness of water measured by the EDTA method is from 250 or 1,000.

[Claim 9] The drink as claimed in any one of Claims 1 to 8, which is characterized in that the taste has been adjusted using sugar or sour flavors.

[Claim 10] A method for producing a drink, which is characterized in that collected seawater is desalted to be separated into water and a concentrate, and then either the concentrate or water-soluble mineral components obtained separately from a seawater concentrate are added to said separated water.

[Detailed the description of the invention]

[0001]

[Field of the Invention]

The present invention relates to a drink that can effectively supplement minerals and a method for producing the same.

[0002]

[Prior Art]

In a diversified contemporary society, consumers' need for foods and drinks has resulted in a great variety of selections. With trends towards health consciousness and orientation to taste as a background, various ideas and improvements have been implemented and a great number of new products for consumers are on the market.

As to drinks, carbonated drinks, coffee drinks, and juices have been widely used for a long time. Saccharide and caffeine, which are main ingredients of those drinks, however, are considered to have rather a harmful effect on health. Consumers are shying away from those drinks. Instead, healthy green tea and sports drinks are gaining in popularity these days. It is expected that trends towards taste orientation and health consciousness will continue to be strong and will be of greater importance in the future. Demand for healthy drinks satisfying consumers' taste orientation and health consciousness is expected to grow in the drink trade.

Of healthy drinks, drinks containing essential minerals such as magnesium, calcium, iron and vitamins, which the Japanese are said not to take sufficiently, are starting to attract consumers' attention for their excellent effects on health.

[0003]

Conventionally, a typical mineral-fortified drink is called mineral water. Mineral water is produced by adding mineral components obtained from ore to drinking water such as tap water. There are many cases regarding mineral water laid open in the patent journal including Japanese Patent Laid-open No.1997-164390 and Japanese Patent Laid-open No.1997-187777.

These drinks, however, are produced by only adding mineral components. Of the minerals essential for the human body, only one or a few types of target components are included. Although essential mineral components such as magnesium and calcium are added, other essential minor mineral components such as iodine and copper are not included. Some products are produced as drinks by adjusting the taste by adding sugar and sour flavors. In either case, no drinks satisfying as healthy drinks with well-balanced mineral components have been obtained.

[0004]

To solve the problem that conventional mineral water does not contain sufficient essential mineral components and essential minor mineral components, methods for using seawater as

an epoch-making and new mineral source were proposed.

Techniques laid open until now include: Japanese Patent Laid-open No.1985-255729 relating to a mineral supplement produced by desalting seawater to reduce the content of NaCl in seawater and sterilizing it; Japanese Patent Laid-open No.1991-77689 relating to ionic water for producing foods and drinks produced by dissolving in water (1) a precipitate generated by condensing a solution obtained by acidifying seawater, adding an alkali-fortified supplement and removing a precipitate generated, and further cooling the solution, and (2) a solid obtained by removing moisture content from the solution obtained after removing the precipitate; Japanese Patent Laid-open No.1993-219921 relating to a drink in which sea water obtained from the deep sea is mixed with tap water; and Japanese Patent Laid-open No.1997-57257 relating to a device for desalting seawater using a reverse osmosis method and adding mineral components by putting the seawater in contact with ore.

Most of these techniques, however, are not suitable for drinks. Those adding seawater cannot avoid the salty taste. If reducing an amount of seawater to be added to suppress the salty taste, only a very small amount of essential minor mineral components is included. In either case, no so-called "delicious" drink has been realized taking advantage of mineral components contained in seawater. There are products using the water of the Dead Sea or Great Salt Lake as a seasoning or an additive for refreshing drinks. The composition of these products is not uniform in all cases because they are natural products. Additionally, it is necessary to dilute the products when used, and adjusting their original taste is difficult. As a result, only a limited number of consumers accept those products.

[0005]

[Problems that the Invention Intends to Solve]

Based on the problems found in conventional techniques, problems which the present invention intends to solve are: to improve mineral drinks which are important as future drinks to meet the needs of health consciousness and taste orientation, to aim to develop a mineral drink containing well-balanced minerals, and to produce a new mineral drink using seawater as a mineral source.

[0006]

[Means to Resolve the Problem]

Having earnestly examined the problems to be solved so as to improve the above-mentioned conventional drinks and to aim to develop a mineral drink containing well-balanced minerals, the inventors of the present invention made it possible to produce a new mineral drink using seawater as a mineral source.

The present invention is characterized in that a drink is produced by adding water-soluble mineral components to the water obtained by desalting seawater.

[0007]

The basic makeup of the invention comprises (1) to (10) in the following:

(1) A drink produced by adding water-soluble mineral components to water obtained by desalting seawater.

(2) The drink, wherein the water-soluble mineral components are obtained from seawater.

(3) The drink, wherein the water-soluble mineral components obtained from seawater are added to water obtained by desalting the same seawater.

(4) The drink, wherein the mineral components obtained from seawater are magnesium and calcium.

(5) The drink, wherein a weight ratio of magnesium to calcium (Mg/Ca) is adjusted to 'one to three' from 'four to one.'

(6) The drink, wherein a weight ratio of magnesium to calcium (Mg/Ca) is adjusted to 'three to one.'

(7) The drink, wherein hardness measured by the EDTA method is from 100 to 3,000.

(8) The drink, wherein hardness measured by the EDTA method is from 250 or 1,000.

(9) The drink, wherein taste has been adjusted using sugar or sour flavors.

(10) A method for producing a drink by separating seawater obtained into water and concentrate, and then adding either of the concentrate or water-soluble mineral components obtained from a separate concentrate to the water obtained by the separation.

[0008]

The details about the present invention are described below.

The present invention relates to an unconventional new drink and a method for producing the same. It realizes a mineral drink containing well-balanced minerals, which can sufficiently supplement a daily intake of minerals essential for health as well as essential minor minerals.

It also realizes a method for producing a new mineral drink using seawater as a rich source of minerals.

The ocean is the origin of all living creatures on earth and a treasure trove of all mineral elements and mineral resources. It accounts for most water resources on earth. Because of its salt content, using seawater for drinking water has been avoided. Exceptionally, seawater is used as drinking water merely in dry regions with extremely low rainfall by treating it to produce fresh water at a high cost. Mineral components in seawater have never been used before except for extracting salt from seawater.

[0009]

The present invention relates to a new drink effectively using mineral components of seawater, which have never been used before. Using water produced by desalting seawater as a raw ingredient, a drink is produced by adding water-soluble mineral components to it.

As seawater used by the present invention, surface water and deep water can be used. Seawater obtained from any depth or in any sea area can be used, but it should be clean seawater without floating objects and toxic substances.

Desalination processing according to the present invention removes sodium chloride contained in large quantities in seawater. Various regular methods are used for this purpose, for which a reverse osmosis membrane method and a distillation concentration method are desirable. By desalinization, water and concentrate are separated. The water obtained is used as a raw ingredient for a drink. By removing sodium chloride from the concentrate obtained using a heating concentration method, etc., a mineral concentrate rich with mineral components such as magnesium and calcium, which are essential for human beings to live, is obtained. By processing this mineral concentrate, overwintered bittern, bittern gypsum, etc. can be obtained. These can also be used as an additive.

As seawater used for the present invention, surface water can also be used. Deep water obtained at the depth of 100 to 10,000m, preferably at the depth of 100 to 1000m, particularly preferably at the depth of 200 to 500m, however, is suitable.

[0010]

Furthermore, a drink can be obtained by adding and mixing water-soluble mineral components to raw ingredient water obtained by desalting seawater. Water-soluble mineral

components can be obtained from ore, but those mentioned above, which are obtained by concentration of seawater, are preferable.

Water-soluble mineral components used in the present invention signifies important minerals such as magnesium, calcium and iron excluding sodium chloride, and essential minor mineral components such as iodine, copper and zinc.

Furthermore in detail, in the present invention, a drink is obtained by dissolving the above-mentioned concentrate of seawater from which sodium chloride has been removed, overwintered bittern, or bittern gypsum, for example, in raw material water obtained by desalting seawater. During this process, essential minor mineral components such as iodine, copper and zinc, which are contained in seawater, incidentally dissolve in the drink.

[0011]

Magnesium and calcium are the most important minerals for the human body. Until now, almost no healthy drinks, in which magnesium is mixed, however, have been available. Calcium is also essential in large quantities for forming and maintaining bones and teeth. The present invention is to add these two minerals as main ingredients. A mixing ratio of these two minerals can be optional. It is, however, preferable to adjust a magnesium-to-calcium (Mg/Ca) weight ratio to 'one to three' from 'four to one.' This is because if a magnesium-to-calcium weight ratio becomes higher than four, the drink becomes difficult to drink due to bitterness. If a calcium -to- magnesium weight ratio becomes larger than three, the taste becomes metallic or hard.

As compared to a recommended daily calcium intake of 600mg, the current average daily intake by the Japanese is reported to be 570mg. As compared to a recommended daily magnesium intake of 300mg, the reported current average daily intake of calcium by the Japanese is 200mg (Source: the Ministry of Welfare). To supplement the shortfall, It is most preferable to adjust a magnesium-to-calcium weight ratio (Mg/Ca) to 'three to one' from standpoints of supplementing calcium and magnesium and satisfying the sense of taste.

[0012]

For drinks, hardness of water is also important and it affects the taste and easiness to drink.

The drink hardness of the present invention is an index for indicating the content of magnesium ions and calcium ions in water and it is expressed by converting it into a calcium carbonate value.

In the present invention, the EDTA method was used to measure the hardness.

Hardness of 100 to 3,000 is preferable for a drink according to the present invention. If the hardness is below 100, the content of important minerals such as magnesium and calcium and essential minor minerals becomes very low. If it is above 3,000, the taste of a drink becomes hard and it becomes difficult to drink. To make the drink easy to drink and to increase an intake of important minerals such as magnesium and calcium, drink hardness of 500 to 1,500 is preferable.

From a standpoint of a drink mainly taken for supplementing minerals, hardness of 1,000 is taste-wise most preferable. Additionally, if used for general cooking such as cooking rice, hardness of 250 is preferable because it can be used for most cooking.

To improve the taste and drinkability of a drink, in the present invention, the taste can of course be adjusted using sugar and sour flavors. By this adjustment, delicate flavors such as a sweet flavor and a salty flavor are added.

Furthermore, the present invention has a new characteristic in a method for producing the above-mentioned drink. A fundamental of the production method is, as mentioned above, to desalt seawater to separate it into water and concentrate, and to add the concentrate obtained or water-soluble mineral components obtained from the concentrate to the water obtained.

[0013]

[Effects]

Because mineral components such as magnesium and calcium, which are added to the water separated from seawater, are a seawater concentrate, minor elements included in seawater can be added sufficiently to a drink obtained according to the present invention. Both water and concentrate obtained from seawater contain essential minor minerals, minor elements and various substances in very small quantities. These components are considered to add delicate flavors and give marvelous effects on health.

Because sodium chloride is mostly removed, drinks do not taste salty and do not have a bad effect by sodium chloride on health. Because moderate seasoning is possible, a drink with characteristics including easy drinkability, improving health and refreshing, which sufficiently satisfy consumers, can be obtained.

Furthermore, fruit juices, vitamins, etc. can be added if necessary. The drink can fulfill its function as a nutritious drink enriched with nutritious ingredients. By adding ingredients with medical effects, it further can fulfill its excellent function for improving health.

Additionally, by specially adding sodium and potassium obtained from seawater, it can be produced as a drink for quenching thirst from perspiration after workouts or sports.

[0014]

[Best Modes for Carrying Out the Invention]

Modes for carrying out the present invention are specifically described below based on embodiments. The present invention, however, is not limited to these embodiments described below.

[Embodiment 1]

Processing of seawater:

Seawater was separated into water and concentrate using a reverse osmosis membrane method. Using a compact tester manufactured by Toray Industries as a reverse osmosis membrane separator, seawater was separated under the following operation conditions:

Operating condition: Liquid temperature at 15°C (59°F)

Pressure: 56MPa

Flow on the water side: 1.9 L/min.

Flow on the concentrated water side: 16 L/min.

Transmissibility of water separated by the above-mentioned operation was as follows:

[0016]

[Table 1]

	Seawater	Water
Transmissibility (μS/cm)	35,000	131

[0017]

By adding overwintered bittern to bittern adjusted by an electrodialysis device, bittern gypsum was obtained. The bittern gypsum content and the minor constituents content were as shown in Table 2 below.

[0018]

[Table 2]

	Calcium salt
Ca (%)	24.2
Fe (ppm)	5.48

Cu (ppm)	0.66
Mn (ppm)	0.37

[0019]

To measure the magnesium salt content, a magnesium salt solution was obtained by desalination by heating the concentrate and separating calcium salt and sodium chloride. The magnesium chloride content and a breakdown of the minor constituent content were as shown in Table 3 below.

[Table 3]

	Calcium salt
Mg (%)	5.18
Fe (ppm)	0.34
Cu (ppm)	0.74
Mn (ppm)	0.25
Zn	1.48
Co	0.01
Mo	0.28

[0020]

Taste test results

Using the above-mentioned raw ingredients, a taste test was conducted. The test results are shown in Table 4 below. The drink hardness and magnesium-to-calcium weight ratios used for the test are as follows:

<Weight ratio of magnesium to calcium>

A = 6 to 1, B = 5 to 1, C = 4 to 1, D = 3 to 1, E = 2 to 1, F = 1 to 1, G = 1 to 2, H = 1 to 3,
I = 1 to 4 and J = 1 to 5

Drink hardness (measured by the EDTA method) was set to the following values:

50, 100, 200, 500, 1000, 1500, 2000, 3000 and 4000

[0021]

[Table 4]

Hardness (EDTA method)	Magnesium-to-Calcium Weight Ratio									
	A	B	C	D	E	F	G	H	I	J
50	○	○	□	□	□	□	□	□	○	○
100	□	□	○	□	□	□	□	○	□	□
200	□	□	○	□	□	□	□	○	□	□
500	□	□	○	□	□	□	□	○	□	□
1,000	□	□	○	□	□	□	□	○	□	□
1,500	×	□	○	□	□	□	□	○	□	□
2,000	×	×	○	○	○	○	○	○	□	□
3,000	×	×	○	○	○	○	○	○	□	—
4,000	×	×	×	×	×	×	×	×	—	—

Note: An evaluation method used is as follows:

Drinks were tested by the sense of taste of five monitors, adult men and women.

□: Most satisfactory, ○: Satisfactory, □: Slightly poor, ×: Poor

[0022]

Evaluation of test results

Drinks hardness with a below 100 were tasty. However, they contained a very small quantity of magnesium, calcium and minor minerals, which was not desirable from a standpoint of supplementing minerals. Drinks hardness with an above 100 and a magnesium-to-calcium weight ratio exceeding four had a bitter taste of magnesium, which was not desirable. Drinks hardness with an above 100 and a calcium-to-magnesium weight ratio exceeding three had a hard taste, which was not desirable.

When drink hardness was higher than 3,000 and a calcium-to-magnesium weight ratio exceeded four, calcium salt stopped dissolving, which was not desirable. When its hardness reached 4,000, a drink was not tasty, which was not desirable.

As a result, it was found that a drink was tasty when its hardness was from 100 to 3,000 and a magnesium-to-calcium weight ratio (Mg/Ca) was 'four to one' to 'one to three'.

Using the above-mentioned raw ingredients, the results of the ICP-MS analysis of a drink prepared with a magnesium-to-calcium weight ratio (Mg/Ca) of 'three to one' and hardness of 1000 are as follows:

[Table 5]

	Hardness 1000
Na	74 ppm
Mg	200 ppm
Ca	71 ppm
K	69 ppm
Fe	7.4 ppb
Zn	34.5 ppb
Cu	24.4 ppb
I	8.8 ppb
P	2.8 ppb
Se	1.9 ppb
Mn	0.1 ppb

[0023]

[Embodiment 2]

Processing of seawater:

Seawater was separated into dilute water and concentrate using an ion-exchange electrodialysis device. An ion-exchange electrodialysis device manufactured by Asahi Glass Company was used. Operation conditions were as follows:

Operating condition: Effective membrane area: 1.72 cm²/sheet

(14 sheets of cation-exchange membranes and 10 sheets of anion-exchange membranes were used.)

Constant current: 8 A

Flow speed: 200 L/H

Fluid temperature: 14°C (57.2°F)

Transmissibility of water separated by the above-mentioned operation was as shown in Table 6 below:

[0024]

[Table 6]

	Seawater	Dilute Water
Transmissibility (μS/cm)	155,000	2,800

[0025]

Calcium and magnesium content:

The concentrate was condensed and desalted, and was separated to sodium chloride and calcium and magnesium salt solution. The content of calcium and magnesium salt solution was as shown in Table 7 below.

(To add to the above, likewise the embodiment 1, minor minerals such as copper and manganese were also included.)

[0026]

[Table 7]

	Unit	Content
Ca	%	0.74
Mg	%	2.80
Fe	ppb	14
Cu	ppb	15
Mn	ppb	0.16
Zn	ppb	742
Mo	ppb	180
I	ppm	1.3

[0027]

Using the water with the transmissibility of $131\mu\text{S}/\text{cm}$ obtained from the above-mentioned reverse osmosis membrane processing and the calcium and magnesium solution of Table 7, a taste test of drinks whose hardness was adjusted to 'a' to 'I' below was conducted. The test results are shown in Table 8 below. The drink hardness values used for the test are as follows:

a = 50, b = 100, c = 200, d = 500, e = 1,000, f = 1,500, g = 2,000, h = 3,000, I = 4,000.

[0028]

[Table 8]

Hardness (EDTA method)	a	b	c	d	e	f	g	h	I
Test Result	□	□	□	□	□	□	○	○	×

Note: An evaluation method used is as follows:

Drinks were tested by the sense of taste of five monitors, adult men and women.

□: Most satisfactory, ○: Satisfactory, ×: Poor

[0029]

Evaluation of test results

When its hardness was below 100, a drink was tasty. However, it contained a very small quantity of magnesium, calcium and minor minerals, which was not desirable from a standpoint of supplementing minerals.

When its hardness was above 4,000, a drink was not tasty, which was not desirable.

As a result, it was found that a drink was tasty when its hardness was from 100 to 3,000 and a magnesium-to-calcium weight ratio was 'four to one' to 'one to three'.

Using the above-mentioned raw ingredients, the results of the ICP-MS analysis of a drink prepared with a magnesium-to-calcium weight ratio (Mg/Ca) of three to one and hardness of 250 are as follows:

[Table 9]

	Hardness 250
Na	19 ppm

Mg	50 ppm
Ca	18 ppm
K	17 ppm
Fe	1.9 ppb
Zn	8.6 ppb
Cu	6.1 ppb
I	2.2 ppb
P	0.7 ppb
Se	0.5 ppb

[0030]

[Embodiment 3]

Refreshing drinks with an *ume* (Japanese apricot) flavor were prepared using the same raw ingredients as used in the embodiment 1, and a taste test was conducted. The test results are shown below.

Drinks used for the test were adjusted to a magnesium-to-calcium weight ratio of three to one and hardness of 1,000.

[Table 10]

Raw Ingredient	Mixture 1	Mixture 2	Mixture 3
<i>Ume</i> (Japanese apricot) extract	2	4	6
Vitamin C	0.05	0.05	0.05
oligosaccharide	5	5	5
Evaluation	○	□	□

[0031]

Note: An evaluation method used is as follows:

Drinks were tested by the sense of taste of five monitors, adult men and women.

□: Most satisfactory, ○: Satisfactory, □: Slightly poor, ×: Poor

[0032]

[Embodiment 4]

By having rats freely take high-cholesterol feeds and a drink adjusted to hardness of 1,000 for four weeks, the total cholesterol level in rats' blood plasma and LDL-Ch were measured. Male Wister rats of 10 weeks (a group of eight) were used as test subjects. The results of the total cholesterol level and LDL-Ch are shown in Table 11 below.

[0033]

[Table 11]

	Total Cholesterol Level	LDL-Ch
High cholesterol feeds	144	38
High cholesterol feeds + drink w/ hardness of 1,000	101	26

[0034]

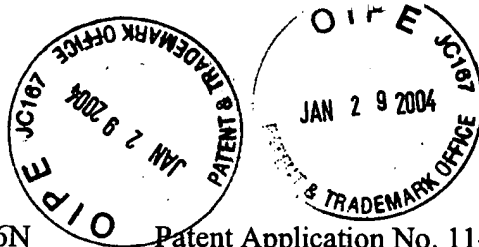
Evaluation of test results

It was confirmed that an increase in total cholesterol and LDL-Ch which is considered to cause arteriosclerosis was suppressed by taking a drink rich in minerals and with hardness of 1,000.

[INDUSTRIAL APPLICABILITY]

As is obvious from the description above, a drink obtained using surface water or deep water according to the present invention can use essential minerals such as magnesium and calcium and essential minor minerals contained in seawater effectively and an amount of magnesium and calcium to be included can be changed optionally.

Because sodium chloride is mostly removed, the drink does not have a salty taste and does not have a bad effect by sodium chloride on health. Additionally, because moderate flavoring of the drink is possible, drinks such as "healthy drinking water", "refreshing drinking water", or others, which are pleasant to drink and which sufficiently satisfy consumers, can be provided.



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[Document Name]

Abstract

[Abstract]

[Problems to be solved]

Heath-promoting drinks sufficiently containing essential minerals and vitamins such as magnesium, calcium and iron, etc.

[Solution]

A method for producing a drink, which is characterized in that collected seawater is desalted to be separated into water and a concentrate, and then adding to the water either one of the concentrate itself or water-soluble mineral components obtained separately from a seawater concentrate.

[FIGURE SELECTED]

None

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